elemental silicon. Steric considerations, the polymer's high solubility, and the propensity of related reactions to give cyclic and polycyclic compounds^{5a,5b,10} suggest a sheetlike or open cage arrangement of fused rings.

While linear polysilanes exhibit strong $\sigma - \sigma^*$ transitions (λ_{max} = 300-350 nm)¹ in the near UV, 1 exhibits a broad and more intense absorption band edge tailing into the visible (Figure 1), associated with extension of Si-Si o-"conjugation" into three dimensions. Poly(n-hexylsilyne) is far more stable to photodegradation in an inert atmosphere than are linear polysilanes (Figure 1), possibly due to the more delocalized nature of excitations or to the greater propensity of a network structure to enforce recombination of photogenerated radicals. Both polydialkylsilanes and 1 photooxidize upon irradiation in air, but while polysilanes fragment to give cyclic oligomers, 1 crosslinks to form polymeric siloxane networks (Figure 2). Also in contrast to linear polysilanes,^{1a} 1 converts directly (without pretreatment) upon pyrolysis to mixtures of Si/SiC without loss of volatile silicon fragments.¹⁰ Further studies on the synthesis and properties of other members of this new class of silicon polymers are in progress.

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Registry No. Na/K. 12056-29-0; hexylSiCl₃ (homopolymer), 113219-09-3.

New Tetrodotoxin Analogues from the Newt Cynops ensicauda

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Tetrodotoxin (TTX, 1), a potent neurotoxin first isolated from puffers1 and then from California newts,2 has recently been reported from various biota.³ Its use as a sodium channel blocker is also expanding. Yet we know little of its biosynthesis or natural analogues.^{3a} Major obstacles are the lack of a detection method for TTX analogues and the poor resolution of ¹H and ¹³C NMR spectra of TTX due to the hemilactal-lactone tautomerism (Figure

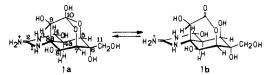
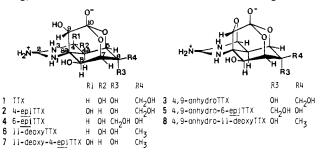


Figure 1. Hemilactal (1a) and lactone (1b) forms of TTX.

1). We previously isolated 4-epiTTX (2) and 4,9-anhydroTTX (3) from puffers.⁴ Both compounds are derivable from TTX with acids and thus provide little information about metabolic pathways. Significantly, we have now isolated from the newt Cynops ensicauda 6-epiTTX, 11-deoxyTTX, and their conversion products.

The newts (3.5 kg) collected in Okinawa, Japan, were extracted with hot 0.1% HOAc, and the extracts were chromatographed successively on columns of charcoal, BioGel P-2, BioRex 70, and Hitachi 3011C ion exchange gel.⁴ Separation of TTX analogues was monitored by a TTX analyzer⁵ and TLC. The following compounds were isolated: TTX (1, 120 mg), 4-epiTTX (2, 15 mg), 4,9-anhydroTTX (3, 20 mg), 6-epiTTX (4, 18 mg), 4,9anhydro-6-epiTTX (5, 3 mg), 11-deoxyTTX (6, 30 mg), 11deoxy-4-epiTTX (7, 2 mg), and 4,9-anhydro-11-deoxyTTX (8, 1 mg). The structural determination of the analogues was



achieved mainly through NMR measurements. Addition of CF₃COOD to the solvent markedly improved the resolution of signals in the NMR spectra of TTX and thus allowed us to assign for the first time all ¹H and ¹³C signals by ¹H-¹H and ¹³C-¹H COSY measurements (Table I).

6-epiTTX (4); $[\alpha]^{25}$ -4.8° (c 0.33, 0.05 N HOAc). The molecular formula by high resolution FABMS⁶ was identical with that of TTX, $C_{11}H_{17}N_3O_8$ (MH⁺, m/z 320.1094, found 320.1106). The LD₅₀ of 4 to mice was 60 μ g/kg (ip). ¹³C and ¹H NMR spectra revealed double sets of signals, thereby indicating that 4 exists as hemilactal-lactone tautomers, as does TTX. The tautomerism was evidenced by negative crosspeaks due to saturation transfer between corresponding protons in a phase sensitive NOESY spectrum. The ratio of hemilactal-lactone tautomers was 6:4. Comparison of ¹³C and ¹H NMR signals of TTX and 6-epiTTX is shown in Table I. Assignments of the signals were derived from ¹H-¹H and ¹³C-¹H COSY measurements. Signals in 4 due to H-4a, H-8, H-11, C-4a, C-5, C-6, and C-7 were significantly shifted from the corresponding signals of TTX, supporting the 6-epi assignment. Proton COSY of 4, showed couplings between H-4/H-4a, H-4a/H-5, H-5/H-7 (W-type), and H-7/H-8, analogous with TTX. The coupling patterns also agreed with those of TTX. Thus structure changes at C-4a, C-5, C-7, and C-8 were ruled out. Because 4,9-anhydro-6-epiTTX (5)⁷ isolated from the same newts was convertible in acid in 4, as is 4,9-anhydroTTX to TTX, C-8a and C-9 of 4 must have the same stereochemistry as in TTX. NOE measurements and difference spectra confirmed the axial substitution of C-11; irradiation at

⁽¹⁰⁾ Bianconi, P. A.; Schilling, F. A.; Weidman, T. W., to be submitted for publication.

⁽¹¹⁾ Sharp, K. G.; Sutor, P. A.; Williams, E. A.; Cargioli, J. D.; Farrar,

 ⁽¹¹⁾ Sharp, K. G., Subi, T. Y., Winanis, E. A., Cargioli, J. D., Farrar, T. C., Ishibitsu, K. J. Am. Chem. Soc. 1976, 98, 1977.
 (12) (a) Michalczyk, M. J.; West, R.; Michl, J. J. Am. Chem. Soc. 1984, 106, 821. (b) West, R.; Fink. M. J.; Michl, J. Science (Washington, DC) 1981, 214, 1343.

⁽¹³⁾ Yokelson, H. B.; Maxka, J.; Siegel, D. A.; West, R. J. Am. Chem. Soc. 1986, 108, 4239.

^{(1) (}a) Tsuda, K.; Ikuma, S.; Kawamura, M.; Tachikawa, R.; Sakai, K.; Tamura, C.; Amakasu, O. Chem. Pharm. Bull. 1964, 12, 1357-1374. (b) Woodward, R. B. Pure Appl. Chem. 1964, 9, 49-74. (c) Goto, T.; Kishi, Y.; Takahashi, S.; Hirata, Y. Tetrahedron 1965, 21, 2059-2088.

<sup>Takahashi, S.; Hirata, Y. Tetrahedron 1965, 21, 2059-2088.
(2) Mosher, H. S.; Fuhrman, F. A.; Buchwald, H. D.; Fischer, H. G. Science (Washington, D. C.) 1964, 144, 1100-1110.
(3) (a) Fuhrman, F. A. Ann. N. Y. Acad. Sci. 1986, 479, 1-14. (b) Yasumoto, T.; Yasumura, D.; Yotsu, M.; Michishita, T.; Endo, A.; Kotaki, Y. Agric. Biol. Chem. 1986, 50, 793-795. (c) Yasumoto, T.; Nagai, H.; Yasumura, D.: Michishita, T.; Endo, A.; Kotaki, Y. Acad. Sci. 1986, 479, 44-51. (d) Yotsu, M.; Yamazaki, T.; Meguro, Y.; Endo, A.; Murata, M.; Naoki H.; Yasumoto, T. Toxicon 1987, 25, 225-228.
(e) Noguchi, T.; Jeon, J. K.; Arakawa, O.: Sugita, H.; Deguchi, Y.; Shida, Y.; Hashimoto, K. J. Biochem. 1986, 99, 311-314.</sup>

⁽⁴⁾ Nakamura, M.; Yasumoto, T. Toxicon 1985, 23, 271-276.

⁽⁵⁾ Yasumoto, T.; Michishita, T. Agric. Biol. Chem. 1985, 49, 3077-3080. (6) The HRFABMS was measured on a JEOL JMS-DX303HF mass

⁽⁶⁾ The HRFABMS was measured on a JEOL JMS-DX303HF mass spectrometer by K. Nojima of JEOL Co. (7) 4,9-anhydro-6-epiTTX (5); $[\alpha]^{25}_{D}$ +10.0° (c 0.08, 0.05 N HOAc); HRFABMS, MH⁺ m/z 302.0999 (calcd for C₁₁H₁₆N₃O₇ 302.0988); ¹H NMR of a hemilactal form δ 5.54 (H-4, s), 2.87 (H-4a, d, $J_{4a-5} = 2.3$ Hz), 4.33 (H-5, dd, $J_{5-4a} = 2.3$, $J_{5-7} = 2.0$ Hz), 4.20 (H-7, t, J_{7-5} , $J_{7-8} = 2.0$ Hz), 4.41 (H-8, d, $J_{8-7} = 2.0$ Hz), 4.61 (H-9, s), 3.65, 3.67 (CH₂-11), d, $J_{gem} = 12.7$ Hz). When irradiated at δ 3.65 (CH₂-11), 12.4% and 13.4% of NOE were observed on H 4a end H 8 competition. observed on H-4a and H-8, respectively

Table I. NMR Spectral Data of TTX, 6-epiTTX, and 11-deoxyTTX^a

	TTX ^b				6-epi-TTX ^c				11-deoxyTTX ^c			
	hemilactal		lactone		hemilactal		lactone		hemilactal		lactone	
	C	Н	С	Н	С	Н	C	Н	С	Н	С	Н
2	156.6		155.9		156.5		155.8		156.4		_	
4	75.1	5.50 (d 9.4)	74.8	5.50 (d 9.4)	75.1	5.55 (d 9.4)	75.1	5.55 (d 8.9)	75.0	5.49 (d 9.4)	74.8	5.51 (d 9.6)
4a	40.7	2.35 (d 9.5)	46.5	2.35 (d 9.5)	41.8	2.01 (d 9.0)	46.9	2.13 (d 9.0)	40.5	2.37 (d 9.4)	46.2	2.37 (d 9.4)
5	73.8	4.25 (br s)	69.2	4.03	75.4	4.30 (d 1.6)	68.4	4.03 (br s)	77.5	4.08 (br s)	72.0	3.87 (br s)
6	71.5	(0. 5)	_d		72.8	(4 110)	77.0	(0. 5)	69.1	(0. 5)	-	(0.))
7	79.7	4.08 (t 1.8)	82.5	4.55 (br s)	82.0	4.08 (br s)	85.5	4.62 (br s)	83.6	3.91 (t 1.6)	86.8	4.35 (t 2.0)
8	72.8	4.30 (d 1.5)	71.5	4.44 (br s)	72.9	4.17 (br s)	71.7	4.26 (br s)	72.6	4.30 (d 1.6)	71.3	4.46 (d 2.3)
8a	59.7		60.4		59.6	. ,	60.1		59.1	, í	59.8	. ,
9	70.9	3.96 (s)	74.0	4.57 (s)	70.8	4.00 (s)	73.7	4.59 (s)	70.8	3.94 (s)	73.9	4.55 (s)
10	110.8		176.1		110.7	.,	175.8		110.6		175.4	.,
11	65.5	4.02 (d 12.6) 4.04	65.2	3.77 (d 12.6) 4.01	65.1	3.74 (s)	66.2	3.68 (d 14.0) 3.69	25.1	1.64 (s)	24.5	1.51 (s)
		(d 12.6)		(d 12.6)				(d 14.0)				

^{a 13}C NMR 75.5 MHz, ¹³CD₃COOD = 22.4 ppm (GN-300); ¹H NMR, CHD₂COOD = 2.06 ppm. Information in parentheses denotes multiplet and J in Hz. ^b1H NMR 360 MHz (NT360) and solvent 1% CF₃COOD, 4% CD₃COOD/D₂O. ^c1H NMR 300 MHz (GN-300), and solvent 4% CD_3COOD/D_2O . ^dUnassignable carbons.

 δ 3.74 (CH₂-11) enhanced signal intensities of H-4a (10.6%) and H-8 (10.3%) of 4, while irradiation of CH_2 -11 of TTX gave no NOE on both protons. All these data support the structural assignment of 4.

The other analogue, 11-deoxyTTX (6), crystallized as colorless needles from 4% HOAc [202 °C dec, $[\alpha]^{25}_{D}$ +5.37° (c 0.34, 0.05 N HOAc)]; the LD₅₀ to mice was 71 μ g/kg (ip); the molecular formula by high resolution FABMS $C_{11}H_{17}N_3O_7$ (MH⁺, m/z304.1145, found 304.1155) corresponded to a monodeoxy derivative of TTX. Proton COSY of 6 showed that the coupling patterns, including a W-type coupling between H-4a and H-9, are essentially the same as those of TTX (Table I), thus suggesting their structural resemblance. Homo and heteronuclear COSY spectra of 6 indicated that the CH_2 -11 signals in TTX were replaced by a methyl signal (Table I). Comparison of ¹³C NMR spectra of 6 and TTX further supported reduction at C-11. A signal assignable to C-6 was shifted upfield, and those of C-5 and C-7 were shifted downfield, while other signals of 6 agreed with those of TTX within 0.6 ppm (Table I). Me-11 was assigned equatorial conformation because no NOE was observed between H-4a and Me-11. These data establish the structure of the new analogue as 11-deoxyTTX, 6. The spectral data indicated that 6 also exists as hemilactal-lactone tautomers (7:3). Two analogues derivable from 6 with acid were also isolated from the newts and identified as 4-epi-11-deoxyTTX (7)8 and 4,9-anhydro-11deoxyTTX (8).9

Biosynthesis of TTX supposedly involves arginine and a C5 unit derived from either amino acids, isoprenoids, shikimates, or branched sugars.¹⁰ The occurrence of 6-epi and 11-deoxy analogues renders branched sugars unlikely precursors, and the shikimate pathway does not seem plausible because it rarely yields 1,2,4-trialkylcyclohexanes. An isoprenoid unit is favored because it possesses both an sp^2 carbon oxidizable to either TTX or 6epiTTX and a methyl that remains in 11-deoxyTTX. The analogues will also be important for structure-activity relationships because chemical transformations of TTX are difficult.

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Models for the Fe^{II}Fe^{III} and Fe^{II}Fe^{II} Forms of Iron-Oxo Proteins

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The importance of redox active, binuclear iron centers in proteins is well established.¹ In an attempt to probe these centers, several crystallographically characterized synthetic analogues of the binuclear iron center in methemerythrin and ribonucleotide reductase have been reported, demonstrating the thermodynamic stability of the $(\mu$ -oxo)bis $(\mu$ -carboxylato)diiron(III) unit.²⁻⁴ Analogues of mixed valence sites in semimethemerythrin and reduced purple acid phosphatases have also been reported, but none are structurally characterized.⁵⁻⁷ In addition, only one model

^{(8) 11-}deoxy-4-epiTTX (7); SIMS (Hitachi M-80) MH⁺ m/z 304; ¹H

^{(8) 11-}deoxy-4-*ep*iTTX (7); SIMS (Hitachi M-80) MH⁺ m/z 304; ¹H NMR of a hemilactal form δ 5.13 (H-4, d, $J_{4-4a} = 4.9$ Hz), 2.87 (H-4a d, $J_{4a-4} = 5.0$ Hz), 4.13 (H-5, d, $J_{5-7} = 1.5$ Hz), 3.91 (H-7, t, J_{7-5} , $J_{7-8} = 1.5$ Hz), 4.29 (H-8, d, $J_{8-7} = 1.5$ Hz), 3.96 (H-9, s), 1.64 (Me-11, s). (9) 4,9-anhydro-11-deoxyTTX (8): SIMS (Hitachi M-80) NH⁺ m/z 286; ¹H NMR of a hemilactal form δ 5.51 (H-4, s), 2.95 (H-4a, d, $J_{4a-5} = 3.0$ Hz), 4.15 (H-5, dd, $J_{5-4a} = 3.0$ Hz, $J_{5-7} = 2.0$ Hz), 4.01 (H-7, t, J_{7-5} , $J_{7-8} = 2.0$ Hz), 4.63 (H-8, d, $J_{8-7} = 2.0$ Hz), 4.56 (H-9, s), 1.61 (Me-11, s). The signals of 5, 7, and 8 were assigned by ¹H⁻¹H COSY measured with a GN-300 spectrometer by using 4% CD₃COOD in D₂O as the solvent. (10) (a) Chevolot, L. In *Marine Natural Products:* Scheuer. P. I. Ed.

^{(10) (}a) Chevolot, L. In Marine Natural Products; Scheuer. P. J., Ed.; New York, 1981; pp 53-91. (b) Shimizu, Y.; Kobayashi, M. Chem. Pharm. Bull. 1983, 31, 3625-3631.

^{(1) (}a) Wilkins, P. C.; Wilkins, R. G. Coord. Chem. Rev. 1987, 79, 195-214. (b) Reichard, P.; Ehrenberg, A. Science (Washington, D.C.) 1983, 221, 514-519. (c) Antanaitis, B. C.; Aisen, P. Adv. Inorg. Biochem. 1983, 5, 111-136.

⁽²⁾ Armstrong, W. H.; Spool, A.; Papaefthymiou, G. C.; Frankel, R. B.; Lippard, S. J. J. Am. Chem. Soc. 1984, 106, 3653-3667.

⁽³⁾ Wieghardt, K.; Pohl, K.; Gebert, W. Angew. Chem., Intl. Ed. Engl. 1983, 22, 727-728

 ⁽⁴⁾ Toftlund, H.; Murray, K. S.; Zwack, P. R.; Taylor, L. F.; Anderson,
 O. P. J. Chem. Soc., Chem. Commun. 1986, 191–193.
 (5) Suzuki, M.; Murata, S.; Uehara, A.; Kida, S. Chem. Lett., Chem. Soc.

Jpn. 1987, 281-284.

⁽⁶⁾ Borovik, A. S.; Murch, B. P.; Que, L., Jr.; Papaefthymiou, V.; Münck, E. J. Am. Chem. Soc. 1987, 109, 7190-7191.

⁽⁷⁾ Hartman, J. R.; Rardin, R. L.; Chaudhuri, P.; Pohl, K.; Wieghardt, K.; Nuber, B.; Weiss, J.; Papaefthymiou, G. C.; Frankel, R. B.; Lippard, S. J. J. Am. Chem. Soc. 1987, 109, 7387-7396.